



Designation: D6722 – 19

Standard Test Method for Total Mercury in Coal and Coal Combustion Residues by Direct Combustion Analysis¹

This standard is issued under the fixed designation D6722; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 These test methods cover procedures to determine the total mercury content in a sample of coal or coal combustion residue.

1.2 *Units*—The values stated in SI units are to be regarded as standard. The values given in parentheses after SI units are provided for information only and are not considered standard.

1.3 **Warning**—Mercury has been designated by EPA and many state agencies as a hazardous material that can cause central nervous system, kidney, and liver damage. Mercury, or its vapor, may be hazardous to health and corrosive to materials. Caution should be taken when handling mercury and mercury-containing products. See the applicable product Material Safety Data Sheet (MSDS) for details and EPA's website (<http://www.epa.gov/mercury/faq.htm>) for additional information. Users should be aware that selling mercury or mercury-containing products, or both, in your state may be prohibited by state law.

1.4 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety, health, and environmental practices and determine the applicability of regulatory limitations prior to use.*

1.5 *This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.*

2. Referenced Documents

2.1 *ASTM Standards*:²

[D121 Terminology of Coal and Coke](#)

¹ This test method is under the jurisdiction of ASTM Committee D05 on Coal and Coke and is the direct responsibility of Subcommittee D05.29 on Major Elements in Ash and Trace Elements of Coal.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

[D2013/D2013M Practice for Preparing Coal Samples for Analysis](#)

[D3173 Test Method for Moisture in the Analysis Sample of Coal and Coke](#)

[D3180 Practice for Calculating Coal and Coke Analyses from As-Determined to Different Bases](#)

[D7448 Practice for Establishing the Competence of Laboratories Using ASTM Procedures in the Sampling and Analysis of Coal and Coke](#)

[D7582 Test Methods for Proximate Analysis of Coal and Coke by Macro Thermogravimetric Analysis](#)

[E691 Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method](#)

3. Terminology

3.1 For definitions of terms used in this standard, refer to Terminology [D121](#).

4. Summary of Test Method

4.1 Controlled heating of the analysis sample in oxygen is used to liberate mercury. The sample is heated to dryness in the instrument and then thermally and chemically decomposed. The decomposition products are carried by flowing oxygen to the catalytic section of the furnace, where oxidation is completed and halogens as well as nitrogen and sulfur oxides are trapped. The remaining decomposition products are carried to a gold amalgamator that selectively traps mercury. After the system is flushed with oxygen to remove any remaining decomposition products, the amalgamator is rapidly heated, releasing mercury vapor. Flowing oxygen carries the mercury vapor through absorbance cells positioned in the light path of a single wavelength atomic absorption spectrophotometer. Absorbance peak height or peak area, as a function of mercury concentration, is measured at 253.7 nm.

4.2 Mercury and mercury salts can be volatilized at low temperatures. Precautions against inadvertent mercury loss should be taken when using this method.

5. Significance and Use

5.1 The emission of mercury during coal combustion can be an environmental concern.

5.2 When representative test portions are analyzed according to this procedure, the total mercury is representative of mass fractions in the sample.

6. Apparatus

6.1 There are several configurations of the instrumental components that can be used satisfactorily for this test method. Functionally, the instrument shall have the following components: drying compartment, decomposition tube, catalyst tube, gold amalgamator, amalgamator furnace, measuring cuvettes, mercury lamp, and detector. The following requirements are specified for all approved instruments (**Note 1**).

NOTE 1—The approval of an instrument with respect to these functions is paramount to this test method, since such approval tacitly provides approval of both the materials and the procedures used with the system to provide these functions.

6.1.1 The instrument shall be capable of drying the sample once it is weighed and introduced.

6.1.2 The instrument shall have a decomposition tube which shall be operated at a temperature high enough to completely decompose the sample. The suggested operating temperature is 800 °C.

6.1.3 The catalyst in the catalytic tube shall be capable of completing the oxidation of the sample and trapping halogens as well as nitrogen and sulfur oxides. The suggested operating temperature of the catalytic tube is 550 °C.

6.1.4 The instrument shall contain a gold amalgamator fixed to an inert material and shall be capable of trapping all mercury.

6.1.5 The amalgamator shall contain a furnace capable of rapidly heating the amalgamator to release all trapped mercury.

6.1.6 The instrument shall have an absorption cell with measuring cuvettes through which the elemental mercury released from the gold amalgamator flows. The cell shall be heated to avoid any condensation of water or other decomposition products.

6.1.7 The light source for the atomic absorption process shall be a low pressure mercury lamp.

6.1.8 A narrow bandpass interference filter or monochromator, capable of isolating the 253.65 nm mercury line, shall be used.

6.1.9 The system may contain a computer for controlling the various operations of the apparatus, for recording data, and for reporting results.

6.2 *Analytical Balance*, with a sensitivity of 0.1 mg.

6.3 *Sample Combustion Boats*, made of nickel and convenient size suitable for use in the instrument being used.

7. Sample

7.1 *Coal*—Prepare the analysis sample in accordance with Practice **D2013/D2013M** by pulverizing the material to pass a 250 µm (No. 60) sieve.

7.2 *Solid Combustion Residue*—Dry a representative portion of the solid residue to constant weight at 110 °C to 115 °C. Determine the moisture loss during this drying step if it is desirable to calculate results to an as-received basis. Crush the dried portion of the sample to pass a 75 µm (No. 200) sieve.

Use a mill that minimizes metal contamination. Use portions of the passing material for analysis.

7.3 *Activated Carbon Sorbent Material*—Activated carbon sorbent material is used in tubes for the absorption of mercury from stack gases. The absorbed mercury is unevenly dispersed in the tubes. The activated carbon material in smaller absorption tubes (100 mg to 200 mg) may be analyzed directly without further preparation. The activated carbon and absorbed mercury in larger sorbent tubes must be homogenized before any sample subdivision. The procedure given in **Annex A2** is recommended for the homogenization and subdivision of the activated carbon sorbent material.³

7.4 Analyze separate test portions for moisture content in accordance with Test Methods **D3173** or **D7582** so that calculation to other bases can be made.

8. Reagents

8.1 *Oxygen*—High purity oxygen, as specified by the instrument manufacturer, shall be used.

8.2 *Certified Reference Materials (CRMs)*—Use Certified Reference Material (CRM) coals with dry-basis mercury values for which confidence limits are issued by a recognized certifying agency such as the National Institute of Standards and Technology (NIST). It is recommended that the user verify the value with the certifying agency before using the CRM coal for quality control purposes.

8.3 All CRMs, reference coals, or calibrating agents must have precision values of less than or equal to method repeatability. Such CRMs, reference coals, or calibrating agents must be stable with respect to moisture and be pulverized to pass 100 % through a 250 µm (No. 60) USA Standard Sieve. CRMs, reference coals, or calibrating agents must be mixed thoroughly before each use.

9. Instrument Preparation

9.1 Assemble the instrumental system in accordance with the manufacturer's instructions. Follow the instrument manufacturer's recommended procedure to optimize the performance of the instrument.

9.2 *Adjustment of Response of Measurement System*—Weigh an appropriate test portion of certified reference material (CRM), calibrating agent, or reference coal. Analyze the test portion (see **9.1**). Repeat this procedure. Adjust instrument response, as recommended by the manufacturer, until the absence of drift is indicated.

9.3 *Calibration*—Select coal CRMs or other calibrating agents and materials specified by the manufacturer that have certified mercury values in the range of samples to be analyzed.

³ This procedure was developed by the LECO Corporation in St. Joseph, MI. The procedure and results from an 11-laboratory interlaboratory study of four analytical methods for the analysis of mercury in Appendix K sorbent tubes can be found in the publication: Evaluation of Methods for Mercury Analysis of Appendix K Sorbent Tubes, EPRI, Palo Alto, CA, Tennessee Valley Authority (TVA), Chattanooga, TN, AEP, Columbus, OH, Consumer's Energy, Jackson, MI, First Energy, Cleveland, OH, LECO Corporation, St. Joseph, MI, Eon U.S., Lexington, KY, Southern Company, Birmingham, AL and TXU Power, Dallas, TX: 2007, 1014565, 110 p.